

Green Solvent-Based Recycling of Lithium from Spent Lithium- ion Batteries






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Why recycling?

Battery Type		Rechargeable	Key Applications
Lithium-Ion (Li-ion)		Yes	EVs, smartphones, laptops, tablets, power tools, grid storage
Lead-Acid		Yes	Car starter, backup power (UPS), solar, forklifts, telecom
Nickel-Cadmium (NiCd)		Yes	Aviation, medical equipment, power tools, railway signaling
Alkaline		No	Remote controls, clocks, flashlights, toys
Zinc-Air		No (mostly)	Hearing aids, medical devices, experimental EVs

Why recycling?

Current state of waste Lithium batteries

Waste LIB sources

Small electronic devices (>80%)

- Laptop
- Cell phone
- Tablet PC
- Camera
- Blue tooth devices



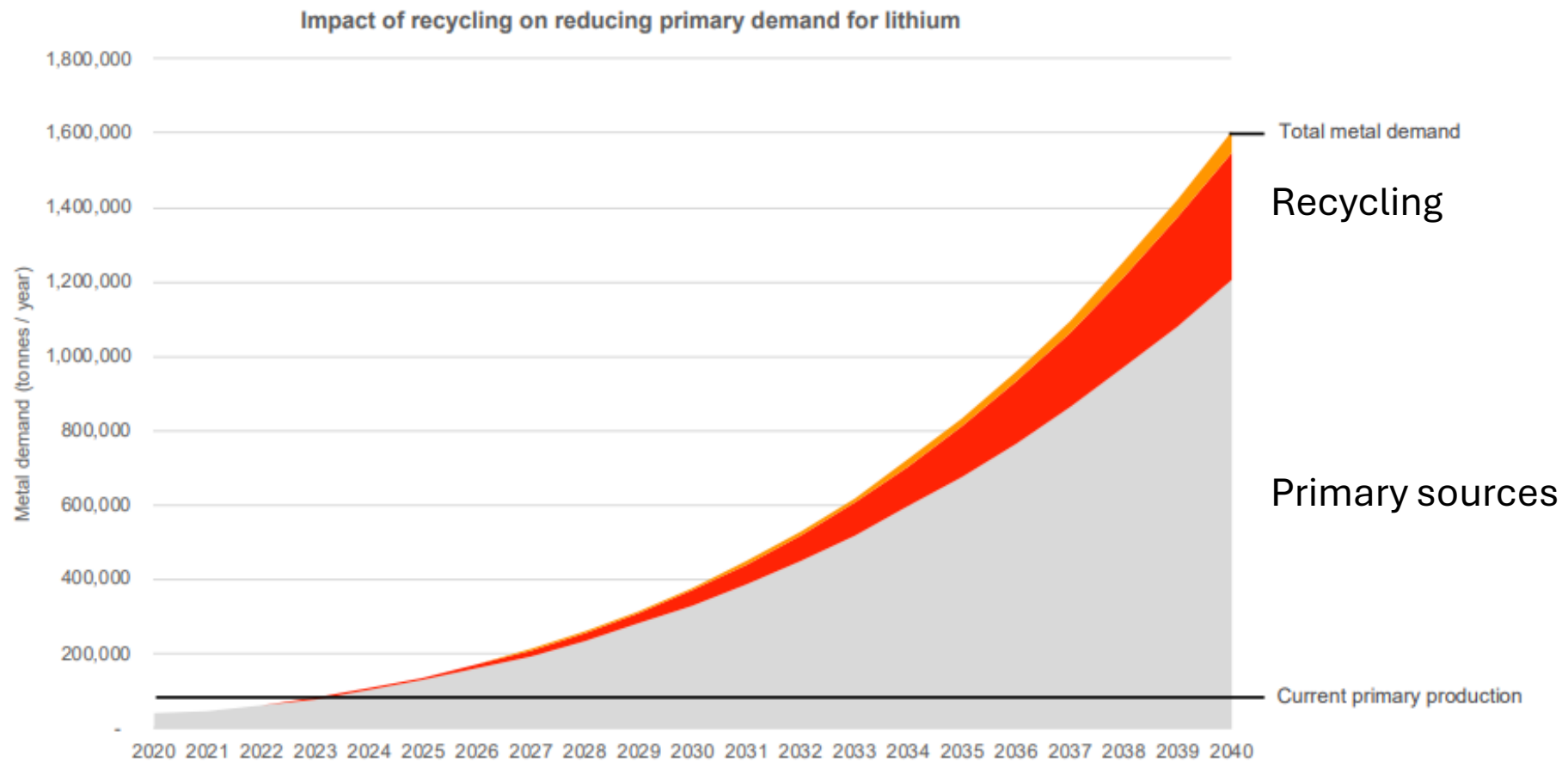
Large electronic devices (< 20%)

- Electric vehicle (EV)
- Energy storage system (ESS)

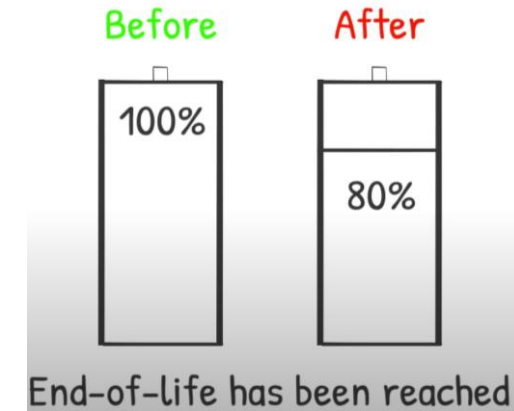
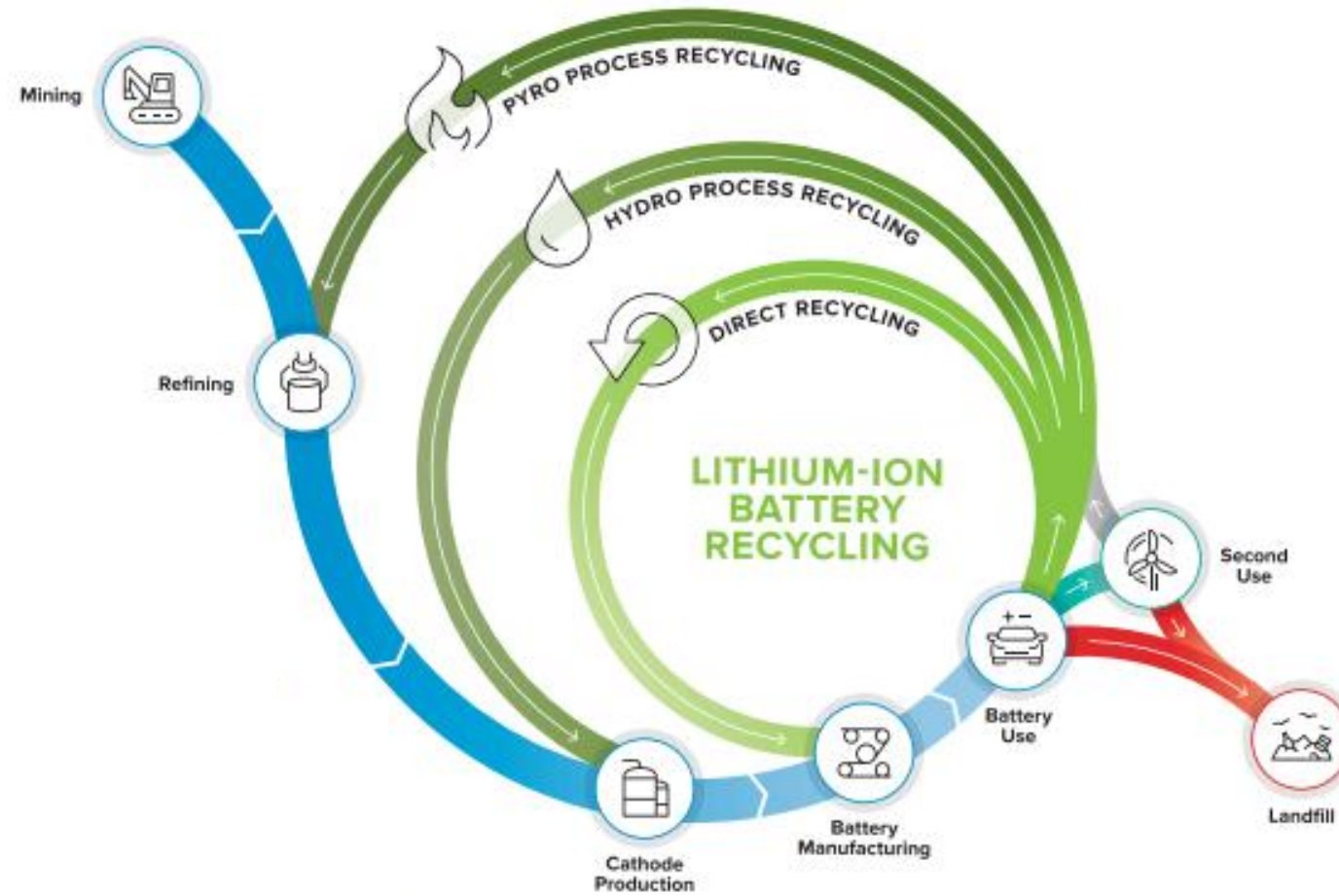


* Small < Li contain 1kg < Large

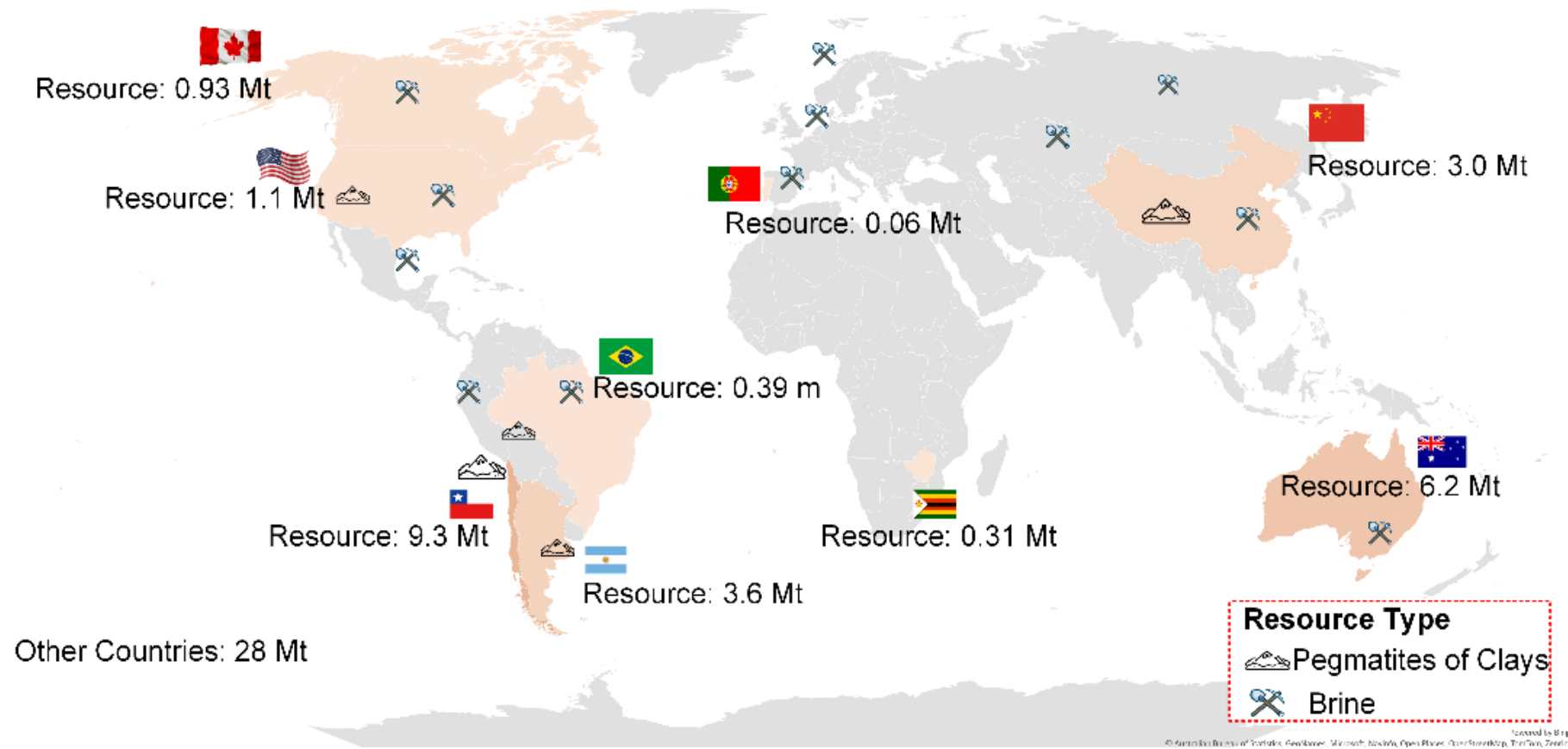
Lithium demands



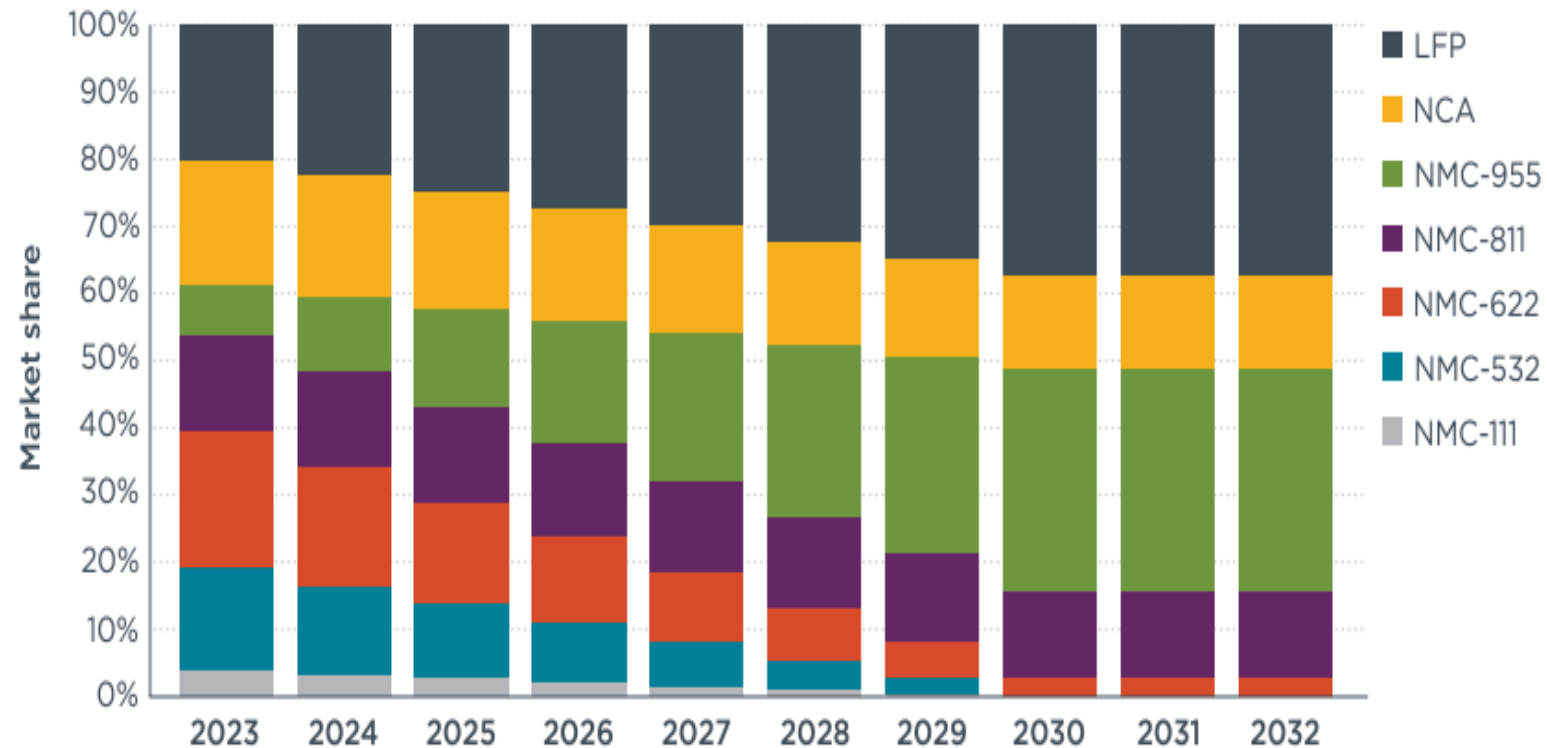
Recycling of LIBs



The lithium landscape



Trends in battery chemistry



Growing demand of LIBs in EVs

Mining vs recycling strategies



Recycling companies

Process name	Feed/Input	Capacity (tonnes/year)	Processing route	Output to battery industry	Output to other industries	Quality of recovered metal
Established LIB recycling processes						
Retriev	LIBs	4500	Mechanical pre-processing / Hydro	metal oxides (incl. CoO)	lithium carbonate (Li ₂ CO ₃), steel, copper, aluminium, cobalt	<i>Recycling</i> – cobalt <i>Downcycling</i> – lithium, nickel, copper
Sumitomo-Sony ('sany process')	LIBs	150	Pyro / Hydro	cobalt (CoO)	cobalt-nickel-iron alloy, copper, aluminium, iron	<i>Recycling</i> – cobalt (processing required) <i>Downcycling</i> – nickel, copper <i>Not recovered</i> – lithium
SungEel HiTech	LIBs	8000	Mechanical pre-processing / Hydro	lithium salts (Li ₂ PO ₄), cobalt (CoO), nickel, manganese	steel, copper, aluminium	<i>Recycling</i> – cobalt, lithium, nickel <i>Downcycling</i> – copper
Recupyl process	LIBs	110	Mechanical pre-processing / Hydro	lithium salts (Li ₂ CO ₃ , LiCO ₂ , Li ₂ PO ₄ , LCO/Co(OH) ₂ /Co)	steel, copper, aluminium, metal oxides (incl. nickel), carbon	<i>Recycling</i> – cobalt, lithium <i>Downcycling</i> – nickel, copper
Umicore process	LIBs (and NiMH bat.)	7000	Pyro / Hydro	cobalt (CoCl ₂), nickel, copper, iron	Slag containing aluminium, silicon, calcium, iron, lithium, manganese, rare earth elements	<i>Recycling</i> – cobalt (ready for LiCoO ₂ synth.), nickel, copper <i>Not recovered</i> – lithium
GEM High-Tech	LIBs	10000	Mechanical pre-processing / Hydro	No data	No data	<i>Recycling</i> – cobalt, nickel ⁶⁰
BRUNP	LIBs	25000	Mechanical pre-processing / Hydro	No data	No data	<i>Recycling</i> – cobalt, nickel ⁶¹
Akkuser process	LIBS	4000	Mechanical pre-processing	Pre-processing only	cobalt, carbon, copper, iron	<i>Recycling</i> – further processing required through hydromet process

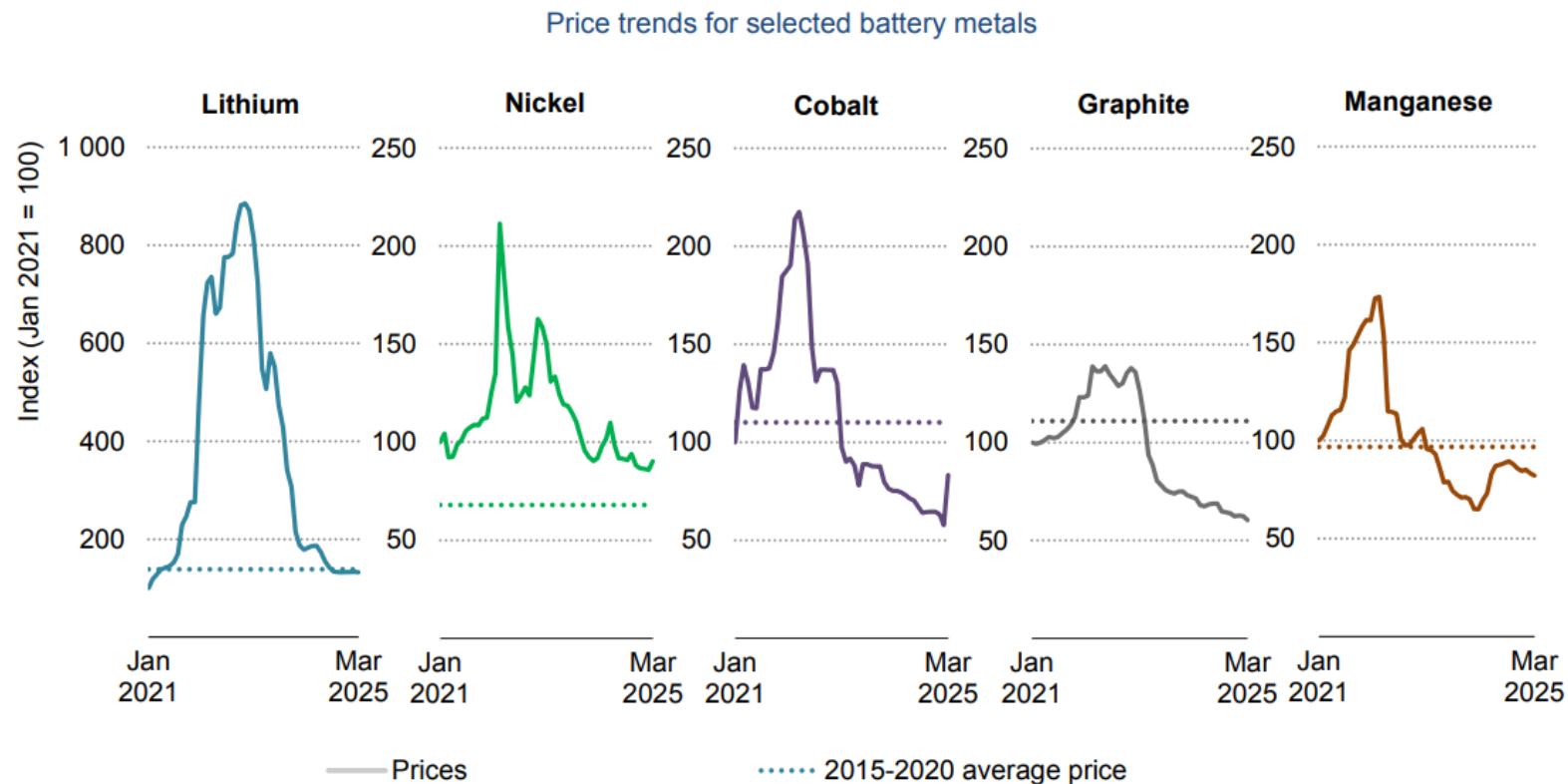
Primary sources vs recycling

Process	Advantages	Disadvantages	Cost	Technological Feasibility	Energy Consumption	GHG Emissions	Water Consumption	Pollutants Generation
Brine Extraction	Lower energy requirement; Low-cost extraction; Abundant in South America (e.g., Salar de Atacama)	Long evaporation time; Vulnerable to climate; Low lithium recovery efficiency	Low to moderate	Mature (but region-specific feasibility)	30,000-36000 MJ/tonne Li_2CO_3	2.7-3.1 tonne CO_2 /tonne Li_2CO_3	Very high (~31-50 m ³ /tonne Li_2CO_3)	Brine depletion; ecosystem disruption
Hard Rock (Spodumene)	High lithium content (~1–2%); Shorter processing time; Less climate dependent	High energy use for calcination; More costly than brine; Mining impacts	High (mining + thermal processing)	Mature (used globally, e.g., Australia)	218,000MJ/tonne Li_2CO_3	20.4 tonne CO_2 /tonne spodumene	High (~77 m ³ /tonne Li_2CO_3)	Tailings, dust, chemical residues
Pyrometallurgy	Simple and scalable; Can treat mixed batteries; Existing infrastructure	High energy demand: Lithium lost in slag; Toxic gas and slag production	High	Mature (commercially applied)	110 MJ/kg battery (CED)	8.81 kg CO_2 -eq/kg (GWP)	Low	SO_x , NO_x , CO_2 , hazardous slag
Hydrometallurgy	High recovery (Li, Co, Ni); Lower energy and GHGs; Selective extraction possible	Requires chemical reagents; Generates wastewater; Multi-step process	Moderate (reagent-dependent)	Mature (lab to industrial scale)	66.87 MJ/kg battery (CED)	3.04 kg CO_2 -eq/kg (GWP)	Moderate	Acidic effluent, heavy metal contamination

Note: GWP: Global warming potential, CED: Cumulative energy demand

Price development

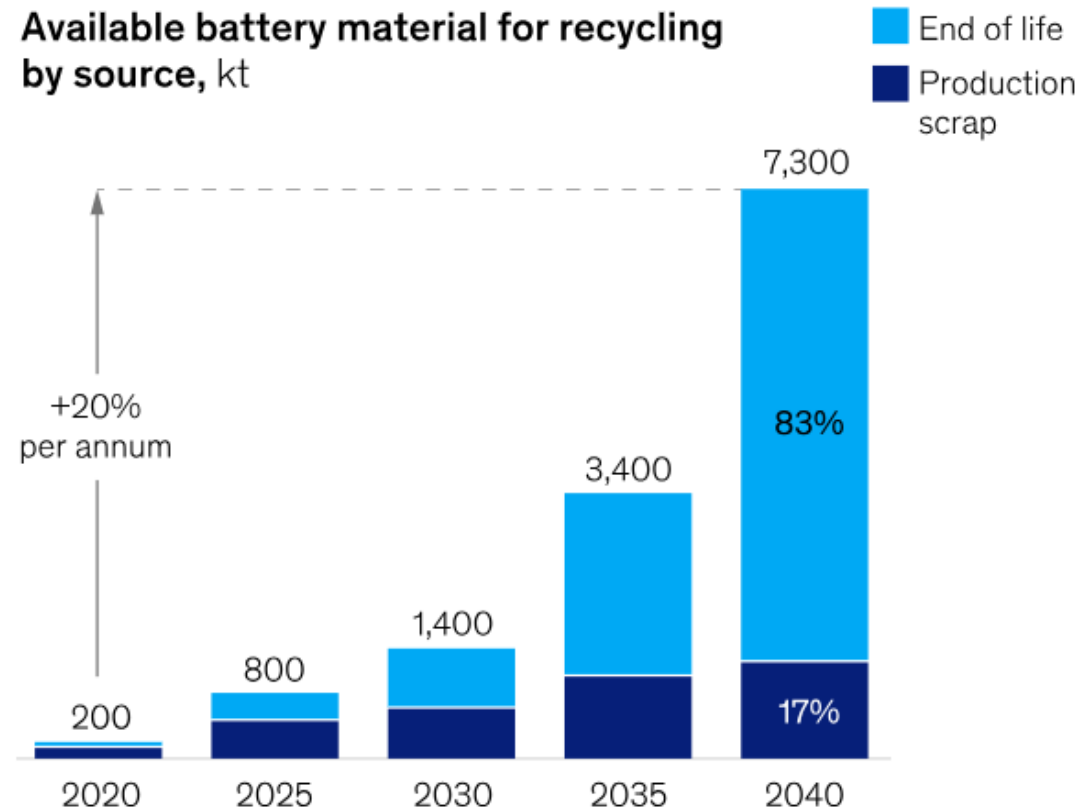
Prices for battery metals continued to decline in 2024 amid growing supply, with the exception of manganese



Recycling cost

USD

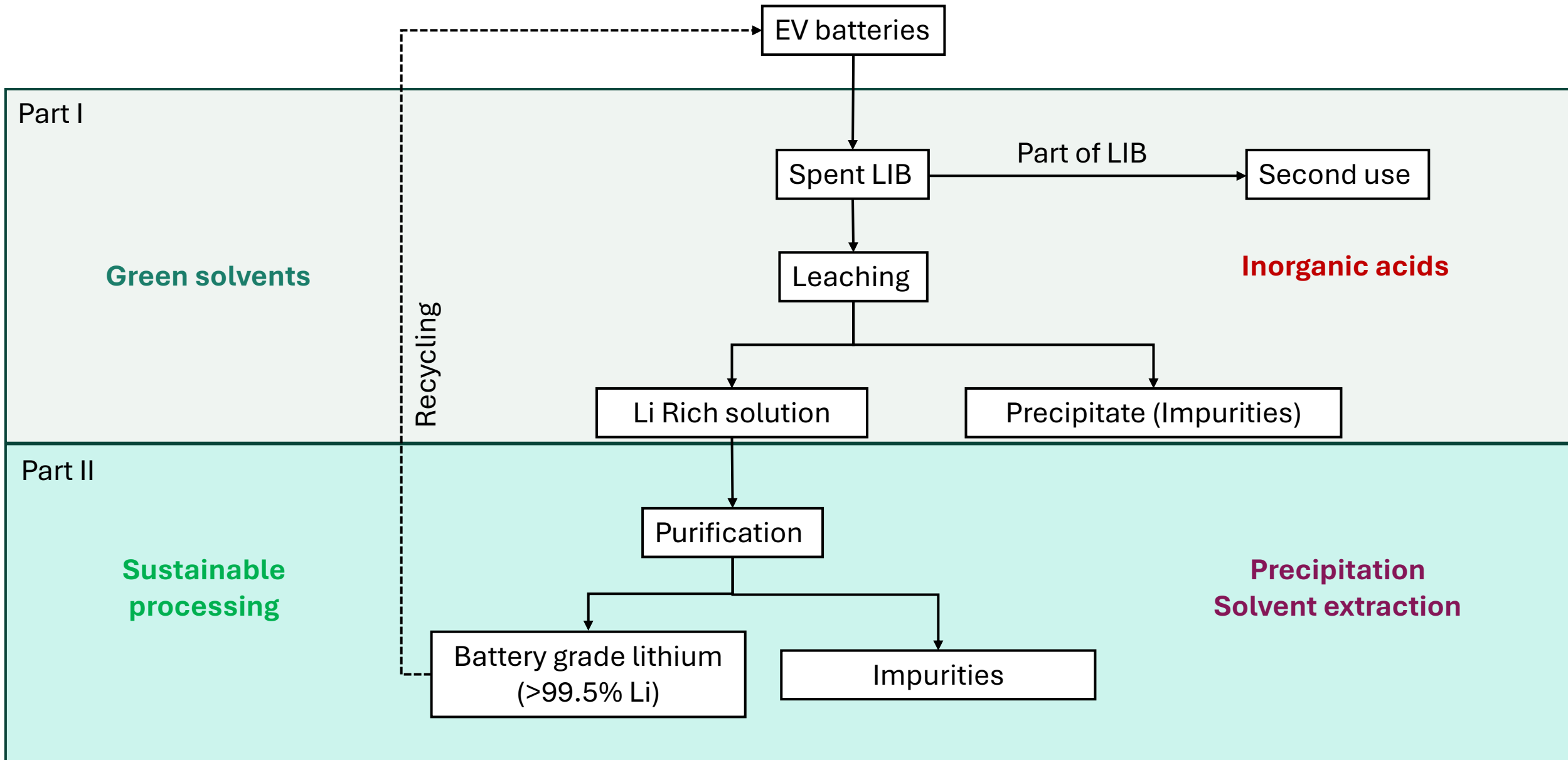
**Available battery material for recycling
by source, kt**



¹Values represent an average across all battery types.

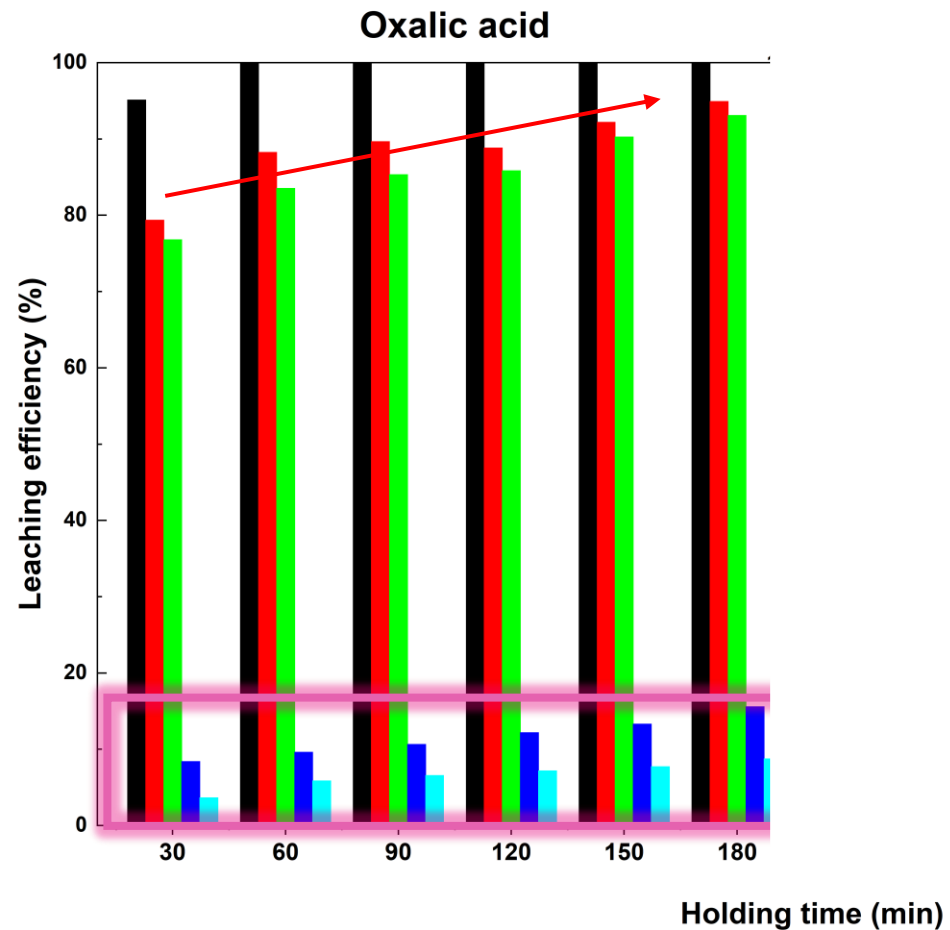
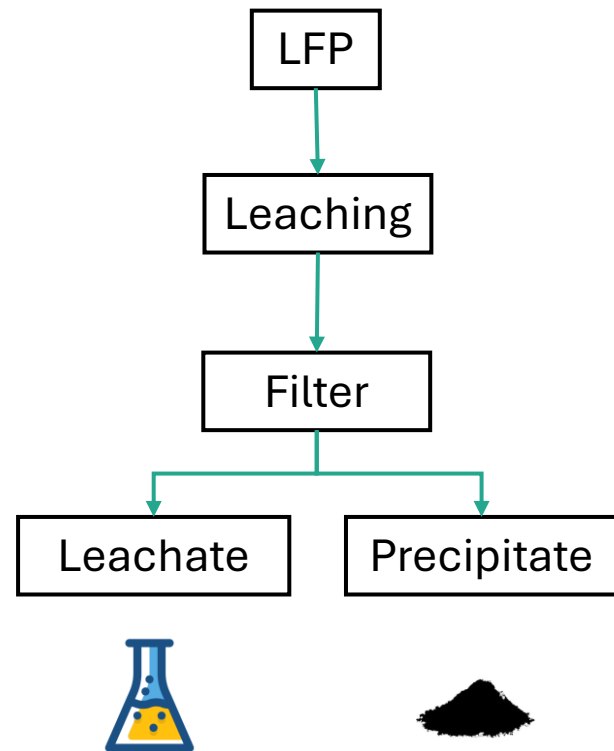
Research approach

Current approach



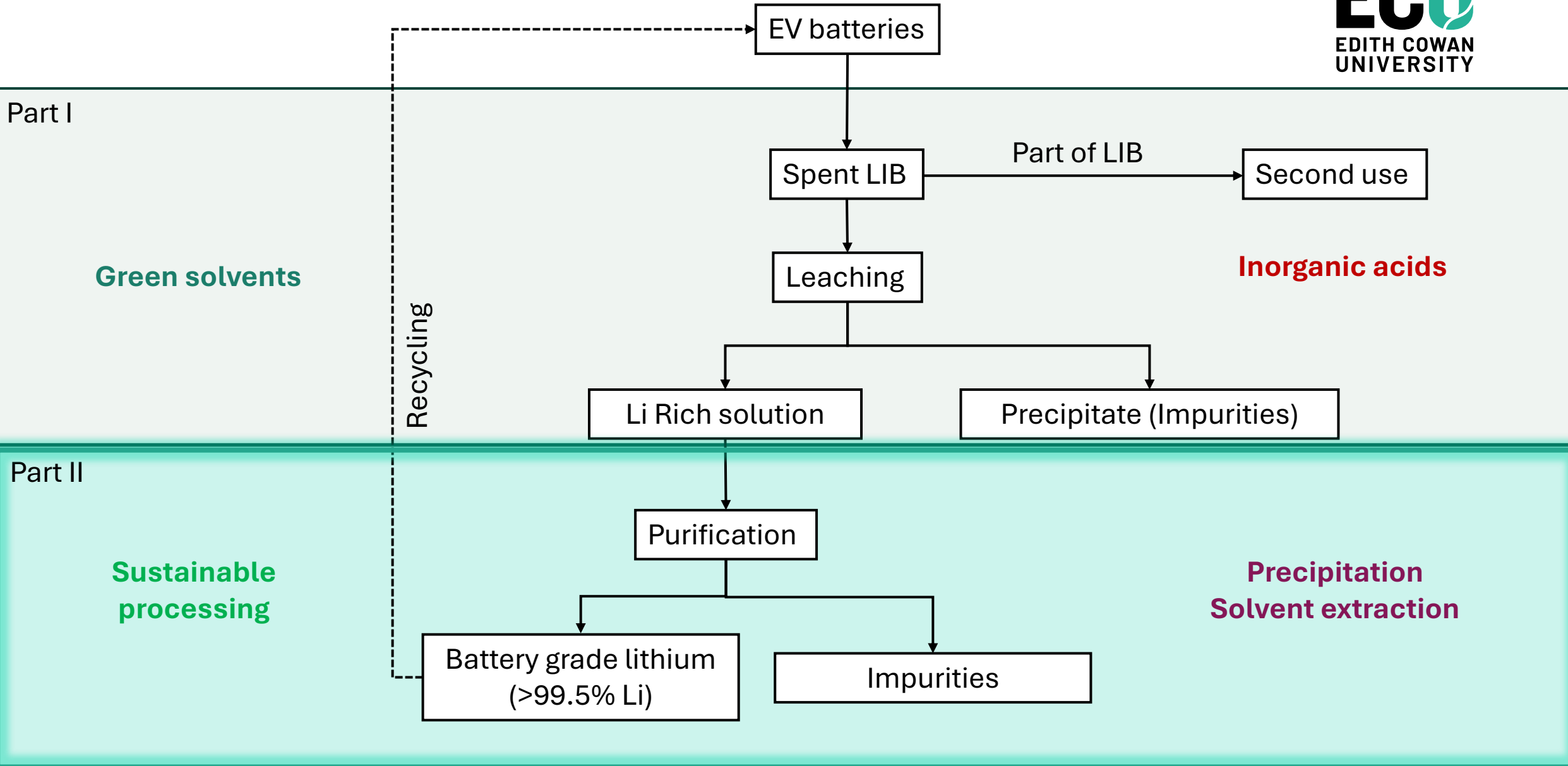
Organic acid leaching

■ P ■ Li ■ Al ■ Fe ■ Cu



Research approach

Current approach



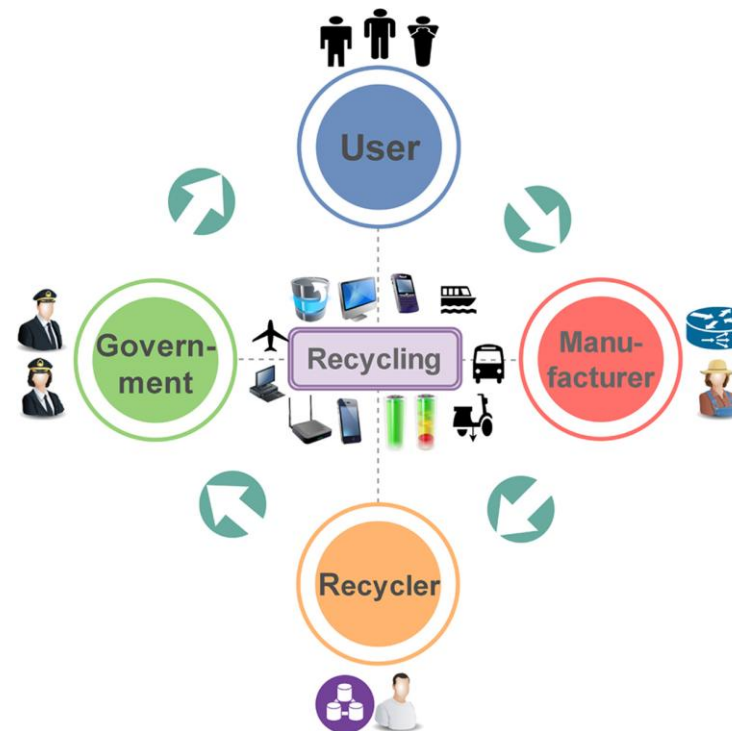
Recycling: Key to sustainable energy transition



Future Perspectives

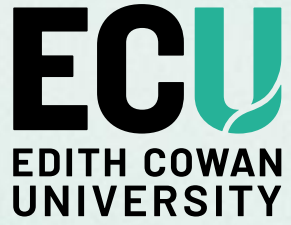
Encourage users to recycle instead of discarding batteries

Provide clear regulations and guidelines for recyclers, and engage the public in the recycling journey



Design batteries that are easy to disassemble and made with less toxic materials and green technologies

Adopt green recycling methods that are safe for both people and the environment



Thank you

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